

# Electricity Lab Wiggle-Bot



Name: \_\_\_\_\_ Date: \_\_\_\_\_

## The Lab

Let's have some fun and learn a little bit about **electricity**. You'll need parts from a disassembled Wiggle-Bot (or Super Wiggle-Bot) to complete this lab.

## Components

Here's what you'll need to complete this part of the activity:



1 - Motor with Holder & Leads



3 - AA Batteries



2 - Dowels  
(each at least 15cm (6") long)



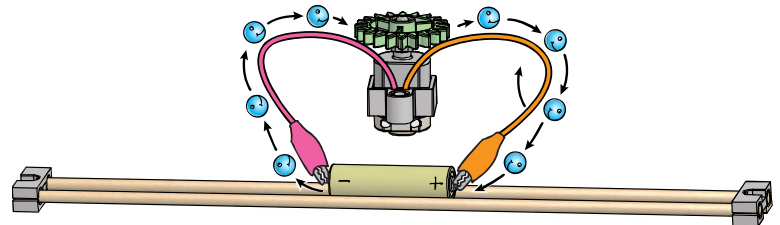
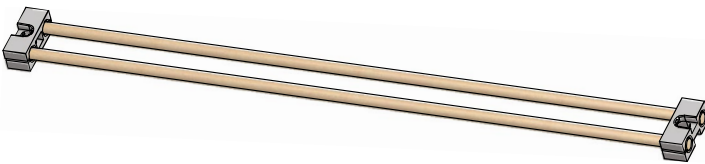
2 - Blocks



Other Materials  
(such as a metal wire, paper clips, connector strips, metal screws, etc.)

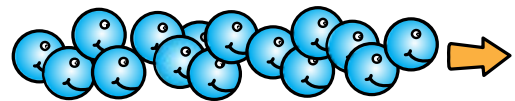
## Let's Get Started

- 1 Make a battery holder out of two **blocks** and two **dowels** as shown below.
- 2 Run your **motor**. Touch your **motor leads** to the ends of a **AA battery**. The **motor** should turn on. Is it magic? Nope, it's **electricity**!



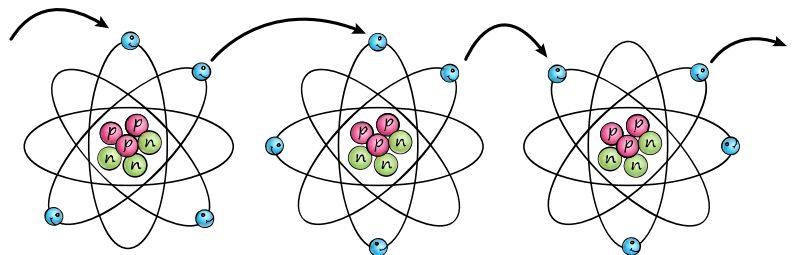
## What is electricity?

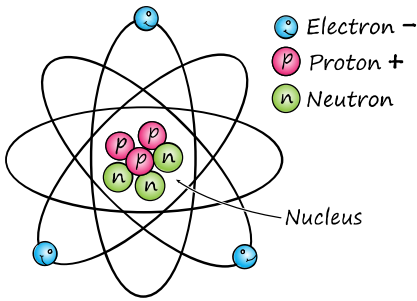
**Electricity** is the flow of **electrons** from one place to another. **Electricity** flows through your motor to make it run.



## Electrons can move?

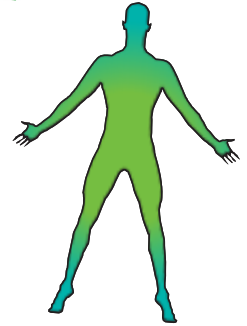
Yes! **Electrons** can move by hopping from **atom** to **atom**. **Electrons** are flowing through you right now, allowing you to think, feel, and move your muscles.



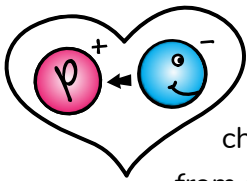


## What is an atom?

Look around you. Every object is made of **atoms**.  
The average adult is made of around  
7,000,000,000,000,000,000,000,000 **atoms**!  
How many **atoms** do you think you are made of?



## Why do electrons move?

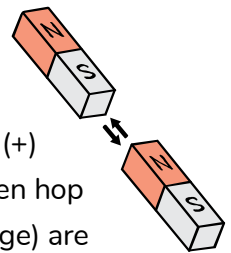


Have you ever heard the phrase “opposites attract”? Well, it’s true.

**Electrons** move because they are attracted to the oppositely charged

**Protons**. **Electrons** have a negative (-) charge and **Protons** have a positive (+)

charge. **Electrons** will do almost anything to move closer to a free **Proton**—even hop from **atom** to **atom**. Only **Electrons** can move. **Protons** and **Neutrons** (neutral charge) are stuck in the nucleus of the **atom**.



1. Fill in the blanks below with the parts of an **atom**.

An \_\_\_\_\_ has a negative (-) charge. A \_\_\_\_\_ has no charge. A \_\_\_\_\_ has a positive charge.

## How does a battery work?

A chemical reaction inside a battery causes the **Electrons** to build up on one side of the battery (the negative side). A battery works because the **Electrons** want to get to the **Protons** on the positive side. However, the **Electrons** cannot travel inside the battery. They need an outside path to get to the **Protons**.

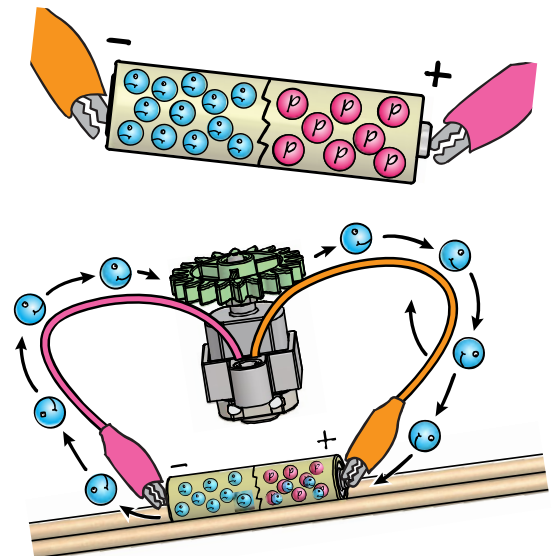
2. What makes the negative side of the battery negative?

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3. What makes the positive side of the battery positive?

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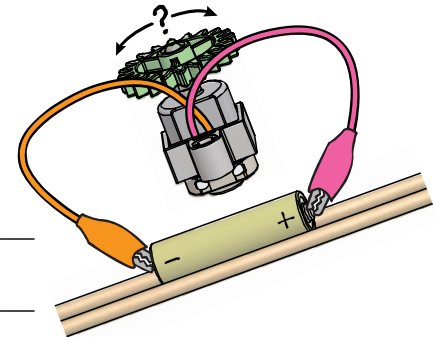


## Where is the reverse?

4. Can you find a way to make your motor turn in the reverse direction?  
Does your solution have anything to do with the direction **electrons** flow?  
Explain how you turned the motor in different directions.

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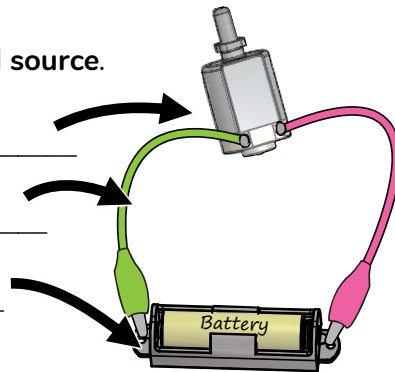


5. Label the parts of the circuit.  
Use the words **load**, **path**, and **source**.

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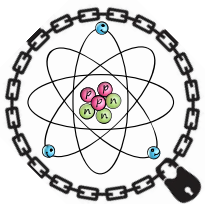
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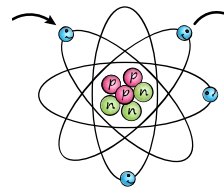
## Circuit

A **circuit** is a complete path for **electricity** to flow. In fact, you created one when you connected both leads of your motor to the battery. In a **circuit**, **electricity** flows from the **source**, through a **load**, and back to the **source**.



## Insulator

**Insulators** are materials that do not let **Electrons** easily flow through them.

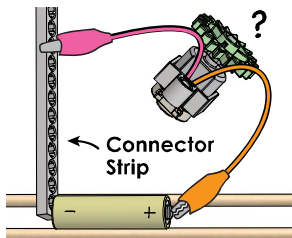


## Conductor

**Conductors** are materials that let **Electrons** easily flow through them.

## What can electricity flow through?

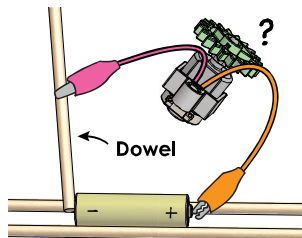
Test different materials to see if they are an **insulator** or a **conductor**. Put them between the battery and one of the motor leads. If the material is a **conductor**, then the **electricity** should flow through it and turn the motor on.



Is it a(n)... (mark your answer)

**Insulator** \_\_\_\_\_

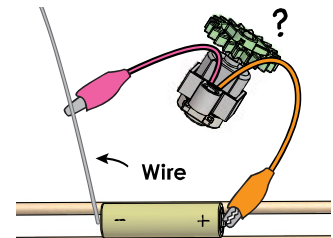
**Conductor** \_\_\_\_\_



Is it a(n)... (mark your answer)

**Insulator** \_\_\_\_\_

**Conductor** \_\_\_\_\_



Is it a(n)... (mark your answer)

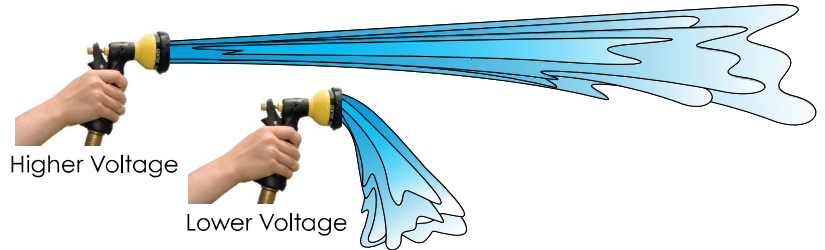
**Insulator** \_\_\_\_\_

**Conductor** \_\_\_\_\_



## What is Voltage (v)?

Voltage is the pressure (or force) that pushes the **Electrons** and causes the flow of **electricity**. It's like water pressure. Look at the picture. The amount of water coming out of the nozzle is the same, but the pressure (or **voltage**) of the water is different. It is measured in **volts**.



6. Look at your AA battery.  
How many **volts** does it produce?

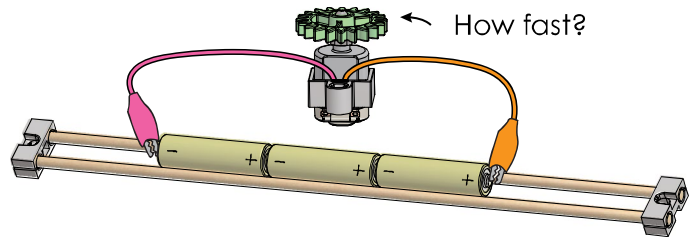
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7. Approximately how many **volts** does a 9-volt battery produce?

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8. Feel the power! Put 2 or 3 batteries together.  
Make sure they are all facing the same way.  
How does this change the speed of the motor?



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9. Create your own electrical experiment. Describe it below. Here are some ideas...

- The graphite (black stuff) in pencils slows down electricity (lowers the **voltage**).  
Can use it to slow down a motor? You will need to make the **electricity** flow through it.
- What happens if you hook up more than one motor to a battery.  
Does it matter how they are hooked up?

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